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34 ChemQuest 11 Name: _____ Date: _____ Hour: _____

Information: Energy Levels and Sublevels As you know, in his solar system model Bohr proposed that electrons are located in energy levels. The current model of the atom isn't as simple as that, however.

~~34 ChemQuest 11 Name: Yumpu~~

Jl ChemQuest 34 Name: Date Hour::F-F Fluorine Gas Molar mass = 38.0 g c.t- Ql Chlorine Gas Molar mass ::Br Br Bromine Liquid Molar mass : I-I' Iodine Solid Molar mass : Information: Some Molecules and Their States Critical

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Thinking Questions What type of force exists between two Fz molecules--{ispersion, dipolar, or hydrogen bonds?,".eh"s#n
?(€ l'5Feq-ece\$'1 What t5.pe of force exists between ...

~~Critical Thinking Questions—LTHS Answers~~

the same exponents. Your answer will them have the same exponent as the numbers you add or. subtract. Here are some examples. Make sure you understand each step:

$4.2 \times 10^6 + 3.1 \times 10^5$ make exponents the same, either a 5 or 6
 $42 \times 10^5 + 3.1 \times 10^5 = 45.1 \times 10^5 = 4.51 \times 10^6$

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32 Information: Light Recall that light is a wave. White light is composed of all the colors of light in the rainbow. All light travels at the same speed (c), 3.00×10^8 m/s. (The speed of all light in a vacuum is always equal to

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(Base the answer to this question on the number of moles of propane that actually get combusted—which is your answer to part a.) 12 moles. For every mole of propane that combusts 3 moles of CO₂ are produced, so the number of moles of CO₂ that can be produced when 4 moles of propane combusts = $4(3) = 12$ ChemQuest 34. Name: _____ Date ...

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A certain element exists as three different isotopes. 24.1% of all isotopes have a mass of 75.23 amu, 48.7% have a mass of 74.61 amu, and 27.2% have a mass of 75.20 amu.

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