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IR PROBLEMS 2

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Answers to IR Spec. Problems - Chemistry LibreTexts

The infra-red spectra of H, I and J are given below, but it does indicate which is - which. Identify the three compounds H, I and J, using the infra-red spectra below, and decide which spectrum belongs to which compound. (iii) Compound E, which is a branched chain haloalkane, was found to have the composition by mass of 39.8% C, 7.3% H, 2)

IR TASK 1 - Weebly

1) 2) 3) 4) 5) 6) bromoethane —¥ ethyl ethanoate cold, aq NaOH nucleophilic substitution propene 2-methylpropylamine chloroethane ethanoic acid

ORGANIC SYNTHESIS PROBLEMS

2-bromo-2-methylbutane 4 5-40 (CH₃ CH₂), 5-40 (CH₃ CH₂), 10-70 (CBr), 5-40 ((CH₃)₂ CBr) methylpropene) 3 90-150 (CH₂), 90 -150 (C₅ 40 H 3 2) propene 3) 90-150 (CH₂), 90 -150 (C₅ 40 H 3 2-chloropropane), 2 5-40 (CH₃ 10-70 (CHCl) propanone), 2 20-50 (CH₃ 190-220 (CO) methylamine 1 25-60 (CH₃)

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2,2-dimethylamine 2-methylbut-1-ene 3-methylbutan-1-ol 4-hydroxybutanone 3-methylbutanoic acid butyl methanoate ethanenitrile propanal Decide which spectrum belongs to which compound and draw the molecule next to the spectrum.

Chemsheets A2 029 (Spectroscopy) - Weebly

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Ir Problems 2 Answers Chemsheets Propene reacts with HBr to form H. H reacts with sodium hydroxide to form I, and I reacts with warm acidified

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potassium dichromate (VI) to form J. The infra-red spectra of H, I and J are given below, but it does indicate which is - which. Identify the three compounds H, I and J, using the infra-

H Nmr Spectroscopy Answers Chemsheets
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1 use your knowledge of ionic equations to give the molar ratio in which the following acids react with bases complete the table to show your answers acid formula of acid base formula of base, www chemsheets co uk 14 jul 12 chemsheets a2 004 1 deduce the rate equation for each of the following reactions a a reacts with b to form c a b c experiment, 3 1 5 kinetics questions by topic aqa ...

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IR PROBLEMS 1 The IR spectra of six compounds are shown. Complete the table to match the spectra to the compounds. Identify any key signals you used to identify each compound. You may not be able to decide between two of the compounds. Structure Spectrum Bond(s) Wavenumber range (cm⁻¹) Structure Spectrum Bond(s) Wavenumber range (cm⁻¹) butanoic ...

IR PROBLEMS 1 The IR spectra of six compounds are shown ...

rate = $k[A][B]^2$ $k = \text{rate} = 16 = 2.0 \text{ mol}^{-2} \text{ dm}^6 \text{ s}^{-1}$ $[A][B]^2$ (2 x 2²) b) rate = $k[E]$ $k = \text{rate} = 0.80 = 0.20 \text{ s}^{-1}$ $[E]$ 4 c) rate = $k[P][Q]$ $k = \text{rate} = 0.00200 = 0.20 \text{ mol}^{-1} \text{ dm}^3 \text{ s}^{-1}$ $[P][Q]$ 0.1 x 0.1 d) rate = $k[H^+][CH_3COCH_3]$ $k = \text{rate} = 9 \times 10^{-5} = 3.0 \times 10^{-5} \text{ mol}^{-1} \text{ dm}^3 \text{ s}^{-1}$ $[H^+][CH_3COCH_3]$ (0.5 x 6.0)

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